

NEW STANDARD ACADEMY

Marks: 80

Date : 21-05-24

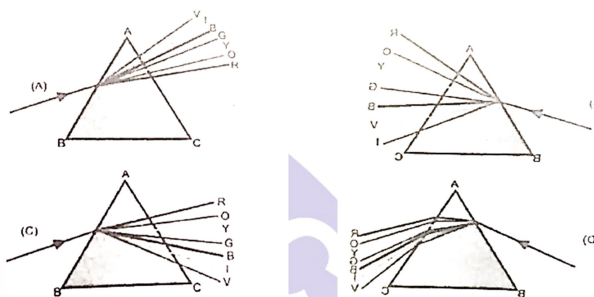
CLASS : 10TH

Time: 90 min.

PHYSICS

- Eye defect at old age is called:
 - Myopia
 - Hypermetropia
 - Presbyopia
 - astigmatism
- When astronauts fly at higher altitude, the sky appears dark because:
 - Scattering of light does not place
 - Scattering of light takes place
 - Refraction of light takes place
 - Dispersion of light takes place.
- When a ray of light passes through a glass prism, it suffers
 - One refraction
 - Two refraction
 - Three refraction
 -) Two reflection
- A person suffering from far – sightedness wears a spectacles having a convex lens of focal length 50 cm. What is the distance of the near point of his eye?
 - 25 cm
 - 30cm
 - 45 cm
 - 50 cm
- In a prism
 - Rays deviate toward the base of the prism.
 - Rays deviate away from base of the prism
 - Rays are reflected internally toward the vertex of the prism.
 - Rays are diffracted around the prism.
- Which colour suffers the least deviation when it passed through a prism?
 - Yellow
 - Red
 - Violet
 - Green
- Four students draw aray diagram showing the dispersion through a glass prism. When a narrow beam of white light incident on the glass prism as shown in figure A,B,C

and D. Which of following is correct ray diagram?



a) A b) B c) C d) D

- Rainbow is formed due to
 - Reflection and dispersion of light through a water droplet
 - Refraction, reflection and dispersion of light through a water droplet
 - Only dispersion of light
 - Only refraction of light
- An object is placed at distance of 15 cm from a convex mirror of focal length 30 cm .Then position of image is-
 - 10 cm
 - 15 cm
 - 20cm
 - 30cm
- An object placed at a distance of 27cm in front of convex lens of focal length 18cm .Then the position of image is
 - 40 cm
 - 54 cm
 - 80 cm
 - 12 cm

CHEMISTRY

- Displacement reaction is:
 - $\text{CaO(s)} + \text{H}_2\text{O(l)} \longrightarrow \text{Ca(OH)}_2\text{(aq)}$
 - $\text{Pb(s)} + \text{CuCl}_2 \longrightarrow \text{PbCl}_2\text{(aq)} + \text{Cu(s)}$
 - $\text{MnO}_2\text{(s)} + 4\text{HCl(l)} \longrightarrow \text{MnCl}_2\text{(s)} + 2\text{H}_2\text{O} + \text{Cl}_2\text{(g)}$
 - $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \longrightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
- The Chemical reaction $\text{HNO}_3 + \text{KOH} \longrightarrow \text{KNO}_3 + \text{H}_2\text{O}$ is an example of:
 - Neutralization

- b) Double Displacement
 - c) Neutralization and double displacement
 - d) Combination reaction
13. Magnesium ribbon is rubbed with sand paper before making it to burn. The reason of rubbing the ribbon is to:
- a) Remove moisture condensed over the surface of ribbon
 - b) Generate heat due to exothermic reaction
 - c) Remove magnesium oxide formed over the surface of magnesium
 - d) Mix silicon from sand paper (Silicon dioxide) with magnesium for lowering ignition temperature of the ribbon

14. The reaction that differs from the rest of the reactions given is:

- a) Formation of calcium oxide from limestone
- b) Formation of aluminium from aluminium oxide. Carbonate
- c) Formation of sodium carbonate from sodium hydrogen carbonate
- d) Formation of mercury from mercuric oxide.

15. When lead nitrate is heated a brown gas is evolved gas is _____

- a) Di oxygen b) Nitrogen Dioxide
- c) Nitrous oxide d) Dinitrogen.

The question given below consist of Assertion and Reason. Use the following key to select the correct answer:

- a) **If both assertion and reason are correct and reason is correct explanation for assertion.**
- b) **If both assertion and reason are correct but reason is not correct explanation for assertion.**
- c) **If assertion is correct but reason is incorrect.**
- d) **If assertion is incorrect but reason is correct.**

16. **Assertion:** In a Chemical reaction, the total mass of the products remains as the total mass of the reactants.

Reason : A chemical reaction involves simply exchange of partners and no new species are formed

17. **Assertion:** Copper can displaced silver from aqueous silver nitrate solution.

Reason: Silver is placed above copper in the reactivity series.

18. **Assertion:** A piece of sodium metal catches fire when throw in water.

Reason: Sodium is a very reactive metal.

19. **Assertion** Chemical combination always takes place between two elements.

Reason : A single substance is formed in a combination reaction.

20. **Assertion :** Quick lime reacts with water to form slaked lime.

Reason: It is a slow chemical reaction.

BIOLOGY

21. Carbonic Anhydrase is present in

- a) WBCs b) RBCs
- c) Platelets d) All of these

22. Which cells have devoid of nucleus

- a) RBCs b) WBCs
- c) Platelets d) None of these

23. Which is known as blood bank –

- a) Liver b) Kidney
- c) Spleen d) Stomach

24. Which is graveyard of RBCs-

- a) Spleen b) Thymus
- c) Bone marrow d) All of these

25. The number of RBCs present in human male

- a) 40-50 lac/mm³
- b) 50-55 lac/mm³
- c) 45 -50 lac/mm³
- d) 30-40 lac/mm³

26. What is the life span of RBCs-

- a) 120 days b) 60 days
- c) 20 days d) 30 days

27. The cells involved in inflammatory reactions are

- a) Basophils b) Neutrophils
- c) Eosinophils d) Lymphocytes.

28. What is the life span of W.B.Cs-

- a) 3-4 days b) 20 days
 c) 30 days d) 40 days
29. Heart is incompletely 4 chambered in:
 a) Fishes b) Amphibians
 c) Reptiles d) Birds
30. Valves are not found in:
 a) Veins b) Arteries
 c) Heart d) All of these

MATHS

31. If the sum of the two roots of the equation $\frac{1}{x+a} + \frac{1}{x+b} = \frac{1}{c}$ is zero, then the product of two roots is ____
- a) 0 b) $\frac{a^2+b^2}{2}$
 c) $\frac{a+b}{2}$ d) $-\frac{(a^2+b^2)}{2}$
32. If 2 is a root of the equation $x^2+bx+12=0$ and the equation $x^2+bx+q=0$ has equal roots, then q =
- a) 8 b) -8
 c) 16 d) -16
33. If roots of the equation $(a^2 + b^2)x^2 - 2(ac + bd)x + (c^2 + d^2) = 0$ are equal then $bc - ad =$ ____
- a) 1 b) 0
 c) -1 d) 2
34. Number of real solution of $(X^2 - 7x + 11)^{x^2-11x+30} = 1$ is
- a) 4 b) 5
 c) 6 d) No solution
35. If $\frac{1}{2}$ is a root of the equation $x^2+kx-\frac{5}{4}=0$ then the value of k is a
- a) 2 b) -2
 c) $\frac{1}{4}$ d) $\frac{1}{2}$
36. If α and β are roots of the eqⁿ $2x^2-5x+3=0$ then the value of $\frac{1}{\alpha} + \frac{1}{\beta}$ is
- a) $\frac{5}{3}$ b) $\frac{3}{5}$
 c) $-\frac{5}{3}$ d) $-\frac{3}{5}$
37. If the equation $2x^2-5x+(k+3)=0$ has equal roots then the value of k is
- a) $\frac{9}{8}$ b) $-\frac{9}{8}$
 c) $\frac{1}{8}$ d) $-\frac{1}{8}$

38. If the system of equation $2x+3y=7$ and $2ax+(a+b)y=28$ represents coincident lines, which of the condition holds true?
- a) $b=2a$ b) $a=2b$
 c) $2a+b=0$ c) $a+2b=0$
39. Two lines with slopes m_1 and m_2 are parallel to each other if:
- a) $m_1=m_2$ b) $m_1m_2=1$
 c) $\frac{m_1}{m_2}=1$ d) $m_1+m_2=1$
40. Find the slope of a line whose inclination with x-axis is 150°
- a) $\frac{1}{2}$ b) $\sqrt{3}$
 c) $-\frac{1}{\sqrt{3}}$ d) None of these